Development of novel freeze-dried mulberry leaves extract-based transfersomal gel

Sopan Nangare1*, Bhatane Dhananjay2, Rushikesh Mali3, Mayuri Shitole4
1* Junior Research Fellow, H. R. Patel Institute of Pharmaceutical Education and Research, Shirpur-425405, Dhule, Maharashtra (India).
2,3,4 Department of Pharmaceutics, Bharati Vidyapeeth College of Pharmacy, Kolhapur-416013, Maharashtra state, India

Abstract:
Nowadays, an antioxidant is important for health associated concern in concert with acne vulgaris. Acne vulgaris is interrelated to the development of free radicals that unite with cells. Mulberry leaves have phenolic compounds that include antioxidants like quercetin. An antioxidant is a scavenger of free radicals. The current study deals with the development of mulberry leaves extract-based transfersomes gel containing quercetin by a thin layer hydration method for topical antioxidant delivery. It was scrutinized by encapsulating the drug in a choice of transfersomes formulations. The batch optimization was carried out by particle size and zeta analysis, entrapment efficiency (%), polydispersity index, in vitro drug release, drug content. An optimized batch MF5 provides entrapment efficiency (%) of quercetin in the vesicles 86.23% and 95.79% drug release. It furnishes vesicle spherical shape with an average diameter of 118.7nm and -45.11mV zeta potential. The MG1 formulation furnishes the superior antioxidant activity, drug content, and entrapment efficiency, ex vivo drug release, spreadability, homogeneity and stability than the MG2. The presence of quercetin in the extract and gel formulation was confirmed by using HPTLC. It is evident from this study that, mulberry leaves extract transfersomes gel are a promising prolonged delivery system for quercetin and have reasonably good stability characteristics. This research recommends that mulberry leaves extract transfersomes gel can be potentially used in the treatment of acne vulgaris through the transdermal drug delivery system.

Keywords: transfersomes, mulberry leaves, quercetin, antioxidant activity, transfersomes gel

1. Introduction
In the modern era, acne is the most common skin disease worldwide (1). The research scenario is around 85% world population affecting between 11 to 30 years at some point. About 4.8 million peoples per year are affected by acne (2, 3). Principally, it is a multifactorial disease of the pilosebaceous unit, which effects in comedones or severe
inflammatory lesions (4, 5) in the skin mainly in the face. This is related to the elevated rate of sebum excretion, abnormal proliferation of keratinocytes, overload production of male hormone androgens, oil-producing glands on the face and inflammatory response initiated by bacterial antigens and cytokines (6). Besides this, oxidative stress is majorly responsible for skin diseases like acne. Principally, the oxidative stress is initiated by free radicals/reactive oxygen species (ROS)(7, 8). In acne vulgaris, the sebum contains hydroxyl, nitrous oxide and superoxide like ROSs are present, which creates irritation during acne, inflammation, etc. Interestingly, the active presence in the plant contains the exceptional antioxidant capacity and the scientific report revealed that the herbal formulation showed the ability to repair damage caused due to the ROS. Additionally, the various scientific reports revealed that the natural antioxidants showed a less adverse effect as compared to the synthetic antioxidant compounds (8). The encapsulation of anti-acne drugs in vesicular and particulate delivery systems is a pioneering and the substitution approach for minimizing the side effects and preserving their efficacy. Presently development of a novel drug delivery system with the intention of high therapeutic activity along with patient compliance with conquering the penetration difficulties associated with transdermal drug delivery system (9). Morus alba L. (mulberry) generally known as a medicinal plant that has been traditionally used. The main content of mulberry leaves extract which is normally referred to phenolics; such as quercetin (quercetin 3-(6-malonylglucoside), isoquercetin, rutin, and some other flavonoids hold have been identified in mulberry leaves(10). It has been used to treat inflammation, cough, hypertension, cancer, and fever due to its medicinal value. The polyphenolic compound of mulberry leaves serves the antioxidant properties by scavenging free radicals and guard many organs against oxidative stress (11). The utilization of mulberry leaves in formulation development can offer several advantages like easy availability, low cost, non-toxic formulation and enhances sericulture farming. Plenty of literature revealed that quercetin reduces the production of interleukin-6 and the expression of metalloproteinase-1 and consequently, reduces inflammation and fibroblast proliferation (12). Also, as an antioxidant, quercetin scavenges the ROS and ultimately repaired the damaged cell. From the past few decades, sustained and efficient drug delivery is taking noteworthy attention of researchers, transdermal drug delivery is one of them and along with that, it offered productive significance and advantages (13). The novel era of drug delivery by using transfersomes exists as a hydrated core surrounded by an ultra-deformable lipid layer complex, which was developed by Gregor Cevc in the year 1991 (14). Generally, transfersomes made up intercellular sealing of lipids. It increases flexibility, reduces the risk of absolute vesicle rupture in the skin and permits transfersomes to penetrate the natural water gradient across the epidermis, after application to the skin. It can be utilized for the delivery of synthetic and herbal drugs (15,16). Recently, considerable attention has been focused on developing a new lipidic nanovesicle based transdermal drug delivery system. Transfersomes have a deformable, flexible and higher affinity for penetration through the skin to the systemic circulation. Transdermal administration of transfersomes vesicles has a great advantage over other vesicles and owing to all these advantages; we attempt to develop the transfersomal gel formulation of mulberry leaves extract. After that, prepared transfersomes gels were evaluated for entrapment efficiency, particle morphology, particle size, zeta potential, polydispersity index, swelling index, viscosity and pH behavior, FTIR, antioxidant activity, deformability index, and penetration study using Franz diffusion cells. HPTLC was used to analyze the level of active substances contained in the extract. At the conclusion, novel transfersomal prepared gel could be exploring the application of natural mulberry leaves extract containing antioxidant (quercetin) for the treatment of acne like skin diseases. Also, it could overcome the drug resistivity and adverse effects of currently engaged acne therapy.

2. Material and methods
2.1. Material
Mulberry leaves (local market, Indapur, Maharashtra), Ethanol AR (70 %) and TWEEN80 (Loba Chemie Pvt. Mumbai), PHOSPHOLIPON 90G, Carbopol 940, PROPYLENE GLYCOL, Formic acid (Merk life sciences), Diethyl ether, Ethyl acetate, Toulene, Petroleum ether, 2,2-diphenyl-1-picrylhydrazyl (DPPH), Ascorbic acid, Quercetin (Yucca enterprises, Mumbai).

2.2. Processing and extraction of mulberry leaves
The mulberry leaves were collected from the local market Indapur, (MS) India, in July. Briefly, the collected leaves were washed systematically under the running water to eradicate soil and other stuff stuck to leaves. Clean leaves were dried, grind and passed through ASTM #30. Powder (50g) of mulberry leaves subjected to Soxhlet extraction using a 70% Ethanol AR as solvent(17). Finally, the extract was subjected to physical and phytochemical examination.

2.3. Freeze drying (FD)
The mulberry leaves extract solution was frozen at -20ºC for 24hr and then dried using lyophilizer (Labconco, United Kingdom.) at pressure 0.013mbar and temperature -49ºC. The obtained extract powder was milled by mortar and pestle to achieve a fine powder.

2.3.1. Isolation of phytoconstituents
FD extract (3g) was successively extracted with 50mL of petroleum ether (fraction I), 50mL of diethyl ether (fraction II) and 50mL of ethyl acetate (fraction III) respectively, with the assist of a separating funnel. Complete extraction was ensured by repeated three times of replication for every batch. Owing to the presence of fatty acids and free flavonoids respectively in fraction I and II was rejected. In addition, fraction III was used for further processing as it contained quercetin, also, it was concentrated and hydrolyzed using 7% sulphuric acid (10mL/g extract) up to 5hr. The hydrolyzed fraction was filtered and extracted with ethyl acetate (1:1/ thrice) via a separating funnel (18).

2.4. Thin-layer chromatography
The isolated portion was chromatographed in comparison with the quercetin as a reference standard (std.) along with Silica gel G TLC plates (Indian herbal pharmacopeia, 2002) by using toluene: ethyl acetate: formic acid (5:4:0.2) as a mobile phase (19).

2.5. UV-Spectrometric analysis
The UV spectrum of the isolated quercetin of mulberry leaves, std quercetin, and the extract was recorded using a UV-Visible Spectrophotometer (Jasco V-630)(20).

2.6. Calibration curve
2.6.1. Calibration curve of std and isolated quercetin
Five working solutions of quercetin with concentrations of 10, 20, 30, 40 and 50μg/mL in methanol were prepared and absorbance was recorded at 246nm λmax for each sample.

2.7. FTIR Spectroscopy
FTIR absorption spectrum of FD mulberry leaves extract was determined by ATR-FTIR (Jasco-V-530 model) using the KBr dispersion method.

2.8. Antioxidant assay
2.8.1. In Vitro antioxidant activity by DPPH method
DPPH is a rapid as well as a sensitive method to estimate the antioxidant capacity of different plant extracts. Also, the visual observation of antioxidant activity is possible. In brief, antioxidants react with DPPH which is a steady free radical and it reduces to the DPPH-H. Reduction in the absorbance of the solution was measured using a UV-Vis spectrophotometer at 517 nm (8, 21). The potential to scavenge the DPPH radical was measured by the following formula(1).

\[
DPPH \ Scavenged(\%) = \frac{Absorbance \ of \ control - Absorbance \ of \ Test}{Absorbance \ of \ control} \times 100 \% \quad \text{(1)}
\]
2.9. Optimization of transfersomes of FD mulberry leaves extract

2.9.1. Preparation of transfersomes

The transfersomes of FD mulberry leaves extract were formulated by the thin layer hydration method (22). The composition of transfersomes for MF1, MF2, MF3, MF4, MF5, MF6 batches reported in Table 1. In these techniques, phospholipon90G (phospholipids), tween 80 (Age activator) and mulberry leaves extract was added to the round bottom flask and dissolved in dichloromethane (solvent). Then, the organic solvent evaporated by using a rotary evaporator above the lipid transition temperature at 50°C, under reduced pressure to velocity used was 60rpm. After complete evaporation of the solvent, a thin layer on the inner wall of the flask was observed. Then, the deposited lipid film hydrated by dehydration process these were performed using phosphate buffer (pH 7.4) until all the thin layer was peeled off. The rehydration process was carried out in the temperature of 37±2°C, 100rpm for 30min. in absence of vacuum condition. Finally, fully hydrated transfersomes suspensions were collected in a vial, then the particle size of the vesicle was reduced by ultrasonication for 5min with the amplitude of 25.

Table 1: Preparation of Mulberry leaves extract transfersomes

2.9.2. Morphological characterization

Prepared transfersomes was subjected to the morphology evaluation. Herein, the Vesicle's morphology of transfersomes was observed by Motic microscope.

2.9.3. Percentage entrapment efficiency (%EE)

The %EE test was performed by using an ultracentrifugation method (indirect method) with 10,000 rpm for 10 min at 4°C, which produces supernatant as a released drug. The total concentration of active compound measurement was performed by dissolving 1ml suspension of transfersomes with PBS pH 6.8 in a 10 ml flask. The concentration measurement was performed by using UV-Visible spectroscopy at 254nm (23). The % EE was calculated by using equation (2).

\[
%EE = \frac{C_{total} - C_{released}}{C_{total}} \times 100
\]

Where,

- \(C_{total}\) = Total concentration of quercetin(μg/ml) in transfersomes.
- \(C_{released}\) = Untrapped concentration of active substances(μg/ml) in transfersomes.

2.9.4. Optimization of transfersomes

The numbers of process variables are accounted for characterization and optimization of the transfersomes formulation. Herein, the batch optimization was carried out by particle size analysis, zeta potential, polydispersity index (PDI), % EE (24).

2.9.5. In vitro drug release of transfersomes

An in vitro drug release study was performed by using modified Franz diffusion cells. Dialysis membrane (Hi-Media, Molecular weight 5000D) was arranged between the receptor and donor compartment. After that, the transfersomes of mulberry leaves extract was kept in the donor compartment and the receptor compartment was filled sufficiently with phosphate buffer, pH 7.4 (25mL). The diffusion cells were maintained at 37±0.5°C temperature with constant stirring at 40rpm throughout the experiment. At the different time (min.) 30, 60, 90, 120, 150, 180, 240, 270, 300 and 360 intervals; 5mL of aliquots were withdrawn from receiver compartment through side tube and again fill with 5mL and analyzed for drug content by UV Visible spectroscopy (25).

2.10. Preparation of transfersomes gel

Herein we used the optimized batch of transfersomes (MF5) for preparation of gel. The 10% (w/v) transfersomes suspension was weighed for gel formation. A carbapol 940 was added in purified water, under stirring and allows hydrating for 24hr. Transfersomes suspension was dispersed in the hydrated Carbapol 940 slurry and stirred continued for 30min. Then, the propylene glycol was added slowly in the slurry. The pH of the formulation is adjusted by
triethanolamine and the same procedure was carried out for control extract based gel (MG2) (Table 2) (26).

Table 2: Transfersomes gel of mulberry leaves extract (MG1) and control extract gel(MG2)

2.11. Evaluation of MG and MG2 gel formulation

MG1 and MG2 gel formulation were subjected to organoleptic evaluation, FTIR, homogeneity rate, pH and viscosity measurements, flow property measurements, drug content analysis (27) and %EE (28).

2.11.1. Homogeneity rate

The determination of homogeneity of the formulated gel was done by pressing the little amount of both gels (MG1 and MG2) among the thumb and the index finger. The uniformity was a resolute as a harmonized or not (29).

2.11.2. Spreadability

The spreadability test of MG1 and MG2 gel was carried out by pressing 0.5g of the final formulation. Briefly, the gel of each batch was subjected between two translucent spherical glass slides and the highest spreading was permitted by leaving them for 5min. The diameter of the formed circle was calculated to articulate the spreadability of the formulated gel (30).

2.11.3. HPTLC analysis

HPTLC analysis was performed by application of the std and isolated quercetin, and MG1 formulation. The samples were prepared by using methanol. After that, the analysis of quertecin was performed on the HPTLC plate of silica gel 60F254 (5cm × 10cm) using a mixture of toluene: ethyl acetate: formic acid as a mobile phase in the proportion of 5:4:0.2. Finally, the quercetin colored bands were identified and confirmed via Rf (31).

2.11.4. LOD and LOQ

Limit of detection (LOD) and limit of quantification (LOQ) are two important terms used to describe the smallest concentration of an analyte that can be reliably measured by an analytical procedure. The LOD and LOQ compounds were detected based on Rf value as well as UV/Visible spectral overlaying of respective standard compounds. The LOD was determined based on the lowest concentration detected by the instrument from each of two standards, while the LOQ was determined based on the lowest concentration quantified in the samples.

The determination of LOD and LOQ was calculated using the formula,

\[ \text{LOD} = \frac{1}{4} \times 3.3 \times (\text{SD}/\text{S}) \]  
\[ \text{LOQ} = \frac{1}{4} \times 10 \times (\text{SD}/\text{S}) \]  

Where,

\( \text{SD} = \text{Standard deviation of the response} \)
\( \text{S} = \text{Slope} \)

2.11.5. Percent of drug content

Accurate quantity of MG1 and MG2 gel was taken in a separate beaker and lysed with 50mL methanol for 15min using ultrasonication and centrifuged at 10,000rpm for 30min (25°C). The clear supernatant was collected and added into the 10mL methanol and diluted with pH 7.4. The drug content was calculated for quercetin an absorbance using a UV spectrophotometer at 246 nm.

2.11.6. Ex vivo release studies

The fresh hairless abdominal skin of the goat was collected from the slaughterhouse and used for release studies after peeling the skin from the underlying cartilage placed inside the receptor compartment (32) (33). In brief, using Franz diffusion cell apparatus the drug release studies were performed. In that, the goat skin was placed in between donor and receptor compartment. The receptor compartment was filled with the phosphate buffer and ethanol mixture in the ratio of 8:2 (15mL) at 37±0.5°C and gel samples were subjected to the donor compartment. Using magnetic stirrer receptor compartment media stirrer at 25rpm, which
avoids the saturation of media during the penetration of the active(34). For calculation of percent ex vivo drug release, the sampling was performed for different time intervals.

2.11.7. Statistical analysis
All outcomes in this work are expressed as a mean ± standard deviation (SD). Paired Student’s t-test was used for the comparison of percent ex vivo drug release of two related gel samples (MG1 and MG2) and assumptions for the statistical significance of p-value. The difference at p < 0.05 was considered to be significant.

2.11.8. Ex vivo penetration test
The ex vivo penetration of transfersomes gel was carried out by the fresh hairless abdominal skin of goat through Franz diffusion cell apparatus assembly. The goat skin was assembled between the donor and receptor compartment with an effective diffusion area of 2.26 cm² and a cell volume of 25 mL. Briefly, the receptor compartment contains the phosphate buffer and ethanol (8:2) media (15 mL) at body temperature 37±0.5°C and stirred the receptor compartment media, the gel was placed in the donor compartment and samplings were carried out periodically for 24 hr from the receptor compartment. Along with that, the sink condition was maintained and the collected samples were subjected to UV spectrophotometer analysis. Based on the experimental finding, the permeation coefficient was calculated by using the cumulative amounts of drug permeated per unit area (μg/cm²) vs time graph. The transdermal flux was calculated from the slope of the linear portion of the graph.

2.12. Stability study
MG1 and MG2 formulations were kept at stability testing for 3 months at a 4°C±2°C and 40°C±2°C / 75%±5% RH stations for determination of the physical and chemical stability of the formulations (as per ICH guidelines).

3. Results of experimental
3.1. Characterization of mulberry leaves extract
The extraction was performed by the successive hot continuous soxhlet extraction method. The ethanolic extract of mulberry leaves showed greenish color. The phytoconstituents analysis was revealed the presence of sterols, tannins, and phenols, alkaloids. The pH of the extract was found in 6.5.

3.1.3. TLC analysis
TLC fingerprinting of mulberry leaves extracts done along with std quercetin under UV 254. It showed that spots with the std quercetin and isolated quercetin having an Rf value 0.38 and 0.31, respectively (Figure 1). At the conclusion, quercetin analysis using TLC pate confirmed that the presence of quercetin in an extract.

Figure 1: TLC plate developed under UV light at 254 nm
After that, the UV spectra of std quercetin was observed at 254 nm, which confirmed the purity and presence of quercetin in the sample prepared sample(Figure 2a). Also, the UV spectra of isolated quercetin from mulberry leaves extract were obtained at 246 nm and confirm the existence of quercetin in extract(Figure 2b).

Figure 2: UV spectra of standard quercetin(a) and isolated quercetin from(b)

3.1.5. Calibration curve of standard quercetin and isolated quercetin
The calibration curve of the std (Figure 3a) and isolated quercetin (Figure 3b) showed the linearity along with 0.999 and 0.996 R², respectively. It provides the confirmation of the purity of quercetin.

Figure 3: Calibration curve of std quercetin(a) and isolated quercetin (b)

3.1.6. FTIR spectra of std and quercetin isolated mulberry extract
FTIR investigations were revealed the presence of quercetin. In brief, std quercetin showed the peak of O-H stretching at 3387.47 cm⁻¹, C=C stretching at 1605.47 cm⁻¹, C-H blending at 1446.70 cm⁻¹, C=O stretching at 1659.83 cm⁻¹, O-H Bending at 1316 cm⁻¹, C-O Aliphatic stretching at 1165.63 cm⁻¹ (Figure 4a). The observed frequencies for isolated quercetin from
the extract of mulberry leaves have been shown in Figure 4b. It showed the O-H (strong) stretching at 3288.26 cm⁻¹, C-H stretching at 2916.94 cm⁻¹, C=O stretching at 1731.51 cm⁻¹, C-H Bending at 1415.10 cm⁻¹, C-O stretching at 1026.95 cm⁻¹ which provides the confirmation of quercetin in the extract (Figure 4b). These pragmatic frequencies confirmed that the isolated fraction was quercetin concerning std quercetin.  

**Figure 4: FTIR spectra of std quercetin(a) and FTIR spectra of isolated quercetin (b)**

### 3.1.7. Antioxidant activity

The antioxidant activity of the FD extract showed an excellent result in contrast to standard ascorbic acid. The isolated quercetin of mulberry leaves, ascorbic acid (std), and std quercetin furnished the antioxidant activity 67.2%, 83.20%, and 69.54% consistently. Among these, isolated quercetin exhibited good free radical scavenging capacity and it’s comparable to the standard quercetin antioxidant activity.

### 3.2. Characterization and optimization of transfersomes

#### 3.2.1. Shape of vesicle

The shape of the vesicle by Motic microscopy was revealed that no aggregation and irregularities in transfersomes, along with that the spherical structures of vesicles were observed in the range of 110 to 460 nm diameter. Hence, it provides confirmation that the successful construction of Transfersomes vesicles (Figure 5).

**Figure 5: Unsonicated transfersomes vesicles by motic microscopy**

#### 3.2.2. Optimization of transfersomes

**Particles size analysis and Polydispersity index (PDI)**

The size distribution of transfersomes suspension was determined by particle size analyzer (Nanoplus) which works on the by photon correlation spectroscopy (Table 3). Owing to the surfactant, the vesicle size observed within the range of 114.5 to 416.6 nm. The PDI values of the formulation were observed in the range of 0.270 to 0.628. Based on the particle size and PDI value, it concludes that the consistent particle size distribution within the formulation.

**Table 3: Characterization and optimization of prepared transfersomes**

**Zeta potential**

The Zeta potential provides the knowledge of particle aggregation or flocculation in suspension. Herein, the transfersomes batch (MF1 to MF6) showed zeta potential in the range of -21.19 to -45.11 mV. These results specify the stability of the transfersomes in a suspension.

**Percent entrapment efficiency**

% EE was calculated by using the ultra-centrifugation method. In brief, the un-entrapped drug was separated and calculated the amount of it. After that, the entrapment efficiency was calculated for transfersomes. From the estimation of the %EE, the MF1 to MF6 showed 57.65 to 86.23 percent %EE.

#### 3.2.3. In vitro release of transfersomes

The determination of percent drug release in phosphate (pH 7.4) was carried out by using the slope of quercetin in phosphate buffer (pH 7.4) calibration curve (Figure 6). The comparison of MF1 to MF6 batch in vitro cumulative release of the diffusion study was shown in Figure 7, it was 90.34%, 92.07%, 91.69% 92.03%, 95.79%, 91.42% respectively. Among all the batches, the MF5 showed the extended-release 95.79% after 6 hrs (Figure 7). Herein, based on the % EE, particle size analysis, zeta potential, and PDI we selected the MF5 batch was an optimized batch and used for further process of gel formulation.

**Figure 6: Calibration curve of extract (pH 7.4)**

**Figure 7: In Vitro release profile of transfersomes**

#### 3.2.4. FTIR spectra of Transfersomes
The FTIR spectra of transfersomes gel showed the peaks for O-H stretching, C-H stretching, C-O stretching and C=O stretching, C-H bending, C-O stretching around 3334 cm\(^{-1}\), 2923 cm\(^{-1}\), 2853 cm\(^{-1}\), 1620 cm\(^{-1}\), 1453 cm\(^{-1}\), 1035 cm\(^{-1}\) respectively, which is an about same as an std quercetin FTIR. The appearance of the above peaks in Figure 8, confirmed the presence of quercetin and other compounds. It was found that there is no interaction between quercetin and excipients used in formulation development.

3.3. Evaluation of MG1 and MG2

The MG2 gel showed a greenish color, glossy appearance. It's because of the direct contact of an extract of the gel-forming agent and MG1 gel furnished slightly greenish color, transparent as well as the glossy appearance. It's because of the extract was entrapped into the lipid vesicles. Both gels showed a somewhat sweet odor and demonstrated exceptional homogeneity.

3.3.1. Measurement of viscosity MG1 and MG2

The viscosity of MG1 and MG2 gel formulations were determined by using by Brookfield Viscometer at different time intervals (Figure 9). MG1 and MG2 showed a remarkable result of viscosity. It is because of carbapol and it could advantageously to resist drug leakage. Concurrently, the pH of the gel was précised by a digital pH meter (Figure 10) and it showed that the MF1 gives the constant pH after 6 hrs and in the case of MF2, variation in the pH was observed.

3.3.2. Swelling index

The MG1 gel showed good water-holding capability. Here in the hydrogen bond facilitates the structure to swell the excipients. A comparison of MG1 and MG2 showed the swelling index up to 99.61 % and 96.27 %, respectively (n=3, Figure 11).

3.3.3. Flow properties

The flow properties of MG1 and MG2 gel were computed at different day intervals. It was observed that the MG1 gel showed better flow properties as compared to the MG2 as shown in Table 4. Outcomes of the flow properties conclude that the MG1 provides exceptional tensile strength, elongation rate, and spreadability. Along with exhibited a good homogeneity rate as compared to the MG2 batch.

3.3.4. Entrapment efficiency, drug content and antioxidant activity

The % drug content of the MG1 and MG2 formulation was found 98.23% and 89.52% respectively (Figure 12a). Besides, the MG1 gel formulation showed superior antioxidant activity (66.72%) as compared to an MG2 gel formulation (59.23%) as shown in Figure 12a. The MG2 gel antioxidant activity, drug content was found to quite less and it can be because of extract getting degraded during the manufacturing process of the gel. The MG1 and MG2 gel formulation showed 85.6%, 81.20% EE. The formation of multi-laminar vesicles complex in transfersomes enhances the %EE (Figure 12b).

3.3.5. HPTLC analysis

Optimized parameter for std quercetin, isolated quercetin and gel formulation (MG1) containing quercetin by HPTLC at 246 nm reported in Table 5. The spectrum scan of std quercetin is comparable with an isolated compound and formulation containing quercetin (Figure 13). For extract, retention time was found to be 8.4 min which coincided with standard quercetin. The results of tests carried on standard and isolated Quercetin and
formulation are summarized in Figure 14. It assured that the presence of quercetin in the gel formulation, along with that there is no interaction between excipients and quercetin.

Table 5: Optimized Parameter of HPTLC for quercetin isolation

Figure 13: a. std. quercetin (Track 1), b. Isolated quercetin (Track 2) and c. gel formulation (Track 3)

Figure 14: Overlay at 246nm of pure quercetin, isolated quercetin, and formulation in HPTLC

3.3.6. LOD and LOQ
The limits of detection (LOD) were determined to be 0.25 and 0.23 ng/spot and the limit of quantification (LOQ) was found to be 0.6 and 0.5 ng/spot for std quercetin and isolated quercetin respectively. The values remain quite similar for both compounds, which revealed the sensitivity of the method.

3.3.7. Ex vivo percent drug release
The ex vivo percent drug release of MG1 and MG2 was carried out in phosphate buffer pH 7.4 through goat skin using Franz diffusion cells (Figure 15). The dissolution profile of MG1 (96.86%) showed an excellent drug release as compared to the MG2 (88.23%), up to 24hrs. A statistically significant test for comparison of the ex vivo release of MG1 and MG2 was done by pairing the T-test. By conventional criteria, the considered difference between MG 1 (Transerosomes gel) and MG 2 (Control gel) was statistically significant at the level of p<0.05.

Figure 15: Dissolution profile of MG2 and MG2 gel

3.3.8. Ex vivo permeation test
The MG1 gel showed better transdermal flux as compared to the MG2 (35.52±3.02 and 26.01±2.02 respectively). Moreover, it has been shown superior permeation coefficient of MG1 (0.016±0.0009) than the MG2 (0.012±0.0003) gel formulation.

3.4. Stability Studies
After 3 months of stability at 4±2°C and 40±2°C in sealed glass ampules confirmed the negligible drug leakage. Based on observation, it concludes MG1 gel formulation was more stables at 4±2°C as compared to the 40±2°C. Furthermore, the leakage of the drug at MG1 batch was found with a minimum as compared to an MG2 gel formulation at both stability stations. The carbapol gel containing viscosity has avoided the movement and fusion of transfersomes; which results in the low drug leakage in gel formulation (MG1). As compared to the drug release of the MG1 gel formulation at zero days after stability, it demonstrated excellent product stability. As a result, there is no major variation seen in MG1, before and after stability (Table 6). Hence, it indicates that the MG1 transfersomes based gel formulation is more stable without causing any incompatibility and it would be a promising potential for topical application.

Table 6: Stability evaluation of transferosomal gel after 3 month

4. Discussion:
Gels are the semi-solid dosage form system of drug delivery and constitute a good repute among novel pharmaceutical dosage forms. Now a day, herbal nano lipid vesicles based gel formulation is becoming admired due to their safe and effective use. In this study, the transfersomes of mulberry leaves extract were prepared by a thin layer hydration method. Numerous investigations have been revealed that quercetin having exceptional antioxidant potential and it decreases the production rate of interleukin-6 and the expression of metalloproteinase-1. Consequently, it reduces inflammation and fibroblast proliferation during the healing process. Besides, plenty of literature claimed that ROS is a major factor for skin diseases (example: acne vulgaris) and antioxidant having scavenging potential to the ROS. The synthetic antioxidant agent having some adverse/side effects. Thus, the nontoxic nature of herbal antioxidants such as quertecin can be effective in the treatment of acne
Herein, the mulberry leaves containing antioxidant active, quercetin was isolated by using TLC, HPTLC, confirmed by using UV spectroscopy, FTIR and compared by using std quercetin. The fruitful outcomes of TLC and UV spectroscopy provide the successful isolation and confirmation of purity of quercetin. The FTIR spectra of std. quercetin, isolated quercetin and gel formulation (MG1) showed the confirmation and excipients compatibility. The development of transfersomes formulation various factors is important. Mainly the size and shape of the vesicle depend on the concentration of phospholipon90G and the concentration of surfactants. An increase in the concentration of surfactant increases the entrapment efficiency of vesicles and ultimately it prevents drug leakage (35,36). Outcomes of experimental tabulated in Table 3, which reveals that the transfersomes prepared by using a specific concentration of tween 80 and phospholipid were shown the superior result as compared to other formulations. Also, the proper proportion of excipients in batch MF5 showed the highest value of %EE (86%). Principally, the particle size of vesicles is an important parameter in the formulation of transfersomes. Generally, particle size distribution based on volume. Moreover, the DLS technique determined PDI value, which is generally in ranges from 0 to 0.6. If the value of PDI > 0.6 indicates that the sample has a very broad size distribution. Result showed that the PDI value of each batch changed with particle size (Table 3). The highest polydispersity index value of MF6 was caused by the large particles which were prone to aggregation (26, 37, 38). Zeta potential is an important parameter to describe the stability of the dispersion system. It is used to measure a magnitude of the electrostatic potential or repulsive force among the same electrical charge of particles in suspension. Also, it gives an idea that particles in suspension undergo aggregation or flocculation. Various scientific reports revealed that the zeta potential is stable when more positive than of +30 mV or more negative than -30 mV (24, 26, 39). Herein, the results revealed that the MF5 transfersomes were more stable than other formulation because their zeta potential is comparatively high. A negative value of zeta potential might be getting from the lipid composition in the formula. Phosphatidylcholine is a zwitterionic compound when isoelectric point 6-7. Also, phosphate buffer, saline at pH 7.4 was used as the hydrating medium in the process of vesicle formulation. In that, the pH was slightly higher than the isoelectric point of phosphatidylcholine and due to this, the phosphatidylcholine carried a negative charge (24). The deformability index is an important parameter in transfersomes preparation and generally, it is used to examine the flexibility of transfersomes. The deformability index value is influenced by the concentration of phospholipid and surfactant used. The use of excessive surfactant can be lower deformability index value as it can lead to the formation of micelles. The larger the index value of deformability, then the transfersomes will be more flexible and allows transfersomes to penetrate through the skin pores smaller than the size transforms itself (40). The result shows that MF5 transfersomes has the lowest deformability index. Based on the transfersomes characterization results batch optimization was carried out and further formulated into a gel formulation. Formula selected was a batch with the highest percentage of drug entrapped, had size distribution, the value of the polydispersity index within range and had the value of zeta potential which more negative than -30 mV. Based on the summary results MF5 transfersomes was chosen because it had a spherical shape, the highest percentage of drug entrapped (86.23±2.1%), polydispersity index of 0.389, zeta potential value of -45, Dv90 particle size of 118.7nm and deformability index of 1.03±0.8. Simultaneously, mulberry leaves extract based gel (MG2) were prepared for evaluation and comparison with MG1. FTIR of MG1 gel confirmed that there is no interaction between quercetin and excipients. Besides, MG1 gel formulation showed satisfactory flow properties, viscosity, and stable pH. Also, MG1 gel showed the superior swelling index and the homogeneity rate. The antioxidant assay of MG1 gel showed that the good antioxidant activity as compared to the MG2. In MG2 formulation, may be chances of
the degradation or interaction between quercetin and excipients and due to that the antioxidant activity get slightly diminished. The HPTLC analysis were used for the further phytochemical and pharmacological investigation in gel formulation. In the present study, the phytochemical constituent (quercetin) was identified and confirmed based on the color zone obtained during the HPTLC analysis. The color of the zone with isolated quercetin and transfersomes gel formulation the color of the reference compound (std quercetin) under daylight and UV light after derivatization in the chromatogram confirmed the presence of stable and pure quercetin. This detailed chemical profile may be useful in the identification as well as quality evaluation of drugs concerning plants. Ex vivo percent drug release results conducted on transfersomes gel and control gel showed a significant difference ($p<0.05$). Furthermore, the penetration rate of MG1 was higher than the MG2. The fruitful findings could be elucidated according to the negative charge of transfersomes (41). These negative charges generated the weak electrostatic repulsion between transfersomes and intercellular components of the skin. After that, this repulsion accelerates the penetration of negatively charged transfersomes through follicles of different skin layers. Additionally, the rapid penetration of transfersomes in the intact part of the basal area of follicles is occurred (42). Herein, the hydration and rehydration temperature, as well as rpm used for the development of transfersomes are key parameters to the penetration mechanism of transfersomes. Also, the phase transition temperature of the surfactant, HLB value of the surfactant, saturation, and unsaturation of alkyl chain length is taking a major account in the penetration mechanism. The prepared M1 gel showed less drug release as compared to control formulation, the possible reason involved in the less leakage from vesicles of transfersomes is greater hydration temperature than the gel to liquid phase transition temperature and along with that, it provides the superior %EE (43, 44). Stability study revealed that the MG1 gel formulation has admirable stability in different stability stations. The outcomes of the present investigation confirmed that the transfersomes of mulberry leaves extract in the gel system offers a new substitute to transdermal drug delivery. Thus, based on the prolific finding the MG1 gel formulation could be utilized in the treatment of acne vulgaris.

**Conclusion**

The present study revealed that the developed novel transferosomal gel of mulberry leaves extract, which enhanced the antioxidant activity as compared to MG2 gel. The optimized batch of transfersomes confirms the excellent zeta potential, particle size, EE (%), PI, deformability index and in vitro percent drug release. Moreover, the MG1 gel ex vivo drug release and penetration studies indicated that developed transfersomes gel formulation may serve as a promising carrier for better penetration through the skin as compared to the MG2. Although, MG1 gel demonstrated good homogeneity, spreadability, excipient compatibility, and drug content. As compared to the MG2, the percent EE and antioxidant properties of MG1 was admirable. Consequently, MG1 batch transfersomes gel of mulberry leaves extract offers tremendous antioxidants potential, which creates new opportunities in the topical application for the treatment of acne vulgaris.

**Acknowledgment**

The authors wish to thank Ms. Neha Desai, Assistant Professor (Ashokrao Mane College of Pharmacy, Peth-Vadgaon) for her technical support.

**Declarations of interest**

The authors report no declarations of interest.

**References**


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<tr>
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<td>Tween 80</td>
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<td>Phosphate buffer (saline pH 7.4)</td>
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n=3, (±): sd

Table 4: The flow properties of MG1 and MG2

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<td>30.25±0.9</td>
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<tr>
<td>Elongation rate (%)</td>
<td>200.21±1.4</td>
<td>282.3±2.5</td>
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<tr>
<td>Spreadability(cm)</td>
<td>9.80±0.9</td>
<td>9.98±0.9</td>
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n=3, (±): sd

Table 5: Optimized Parameter of HPTLC for quercetin isolation

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<th>Parameters</th>
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<tbody>
<tr>
<td>Stationary Phase</td>
<td>Merck Silica gel 60 F254 HPTLC pre-coated plates</td>
<td>Plate size</td>
<td>4.0cm x 10.0cm</td>
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<tr>
<td>Mode of separation</td>
<td>Normal phase</td>
<td>Development chamber</td>
<td>Camag twin trough chamber</td>
</tr>
<tr>
<td>Mobile phase</td>
<td>Ethyl acetate: Toluene: Formic acid (4:3.5:0.5v/v/v)</td>
<td>Bandwidth</td>
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<tr>
<td>Chamber saturation</td>
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<td>Space between the bands</td>
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</tr>
<tr>
<td>Sample applicator</td>
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<td>Syringe</td>
<td>Hamilton, 100.0µL</td>
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<tr>
<td>Distance from the edges of the plat</td>
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<td>Rate of a sample application</td>
<td>150nL/sec</td>
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<tr>
<td>Lamp and</td>
<td>Deuterium, 246nm</td>
<td>Development</td>
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uncorrected proof
Densitometric scanner: Camag Scanner IV equipped with win-CATS Planar Chromatography manager software version 1.4.7

Table 6: Stability evaluation of transferosomal gel after 3 months

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<tr>
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<th>MG2</th>
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<tr>
<td>(4°C ±2°C)</td>
<td>a) Color</td>
<td>Slightly yellowish</td>
<td>Greenish</td>
</tr>
<tr>
<td></td>
<td>b) pH</td>
<td>5.62±0.23</td>
<td>5.89±0.95</td>
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<tr>
<td></td>
<td>c) Viscosity(CPS)</td>
<td>14005±230</td>
<td>13202±456</td>
</tr>
<tr>
<td></td>
<td>d) EE (%)</td>
<td>84.96±2.35</td>
<td>78.35±3.69</td>
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<tr>
<td></td>
<td>e) Drug content (%)</td>
<td>98.01±2.61</td>
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Legend

Figure 1: TLC plate developed under UV light at 254 nm
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Figure 15: Dissolution profile of MG2 and MG2 gel.

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Phosphahte buffer (pH 7.4)

\[ y = 0.0299x + 0.0033 \]

\[ R^2 = 0.982 \]

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**Table**

<table>
<thead>
<tr>
<th>Table 1: Preparation of Mulberry leaves extract transfersomes</th>
</tr>
</thead>
<tbody>
<tr>
<td>Material name</td>
</tr>
<tr>
<td>Mulberry leaves extract</td>
</tr>
<tr>
<td>Tween 80</td>
</tr>
<tr>
<td>Phospholipon 90G</td>
</tr>
<tr>
<td>Phosphate buffer (saline pH 7.4)</td>
</tr>
</tbody>
</table>

**Table 2: Transfersomes gel of mulberry leaf extract (MG1) and control extract gel (MG2)**

<table>
<thead>
<tr>
<th>Composition</th>
<th>Transfersomes</th>
<th>Extract</th>
<th>Carbopol 940</th>
<th>Triethanolamine</th>
<th>Propylene glycol</th>
<th>Water</th>
</tr>
</thead>
<tbody>
<tr>
<td>MG1 Equal to 10% extract</td>
<td>-</td>
<td>1</td>
<td>0.1</td>
<td></td>
<td>12.5</td>
<td>Add 100</td>
</tr>
<tr>
<td>MG2</td>
<td>-</td>
<td>10</td>
<td>1</td>
<td>0.1</td>
<td>12.5</td>
<td>Add 100</td>
</tr>
</tbody>
</table>

**Table 3: Characterization and optimization of prepared transfersomes**

<table>
<thead>
<tr>
<th>Batch no.</th>
<th>Particle size</th>
<th>Zeta potential</th>
<th>PDI</th>
<th>% EE</th>
<th>Deformability Index</th>
</tr>
</thead>
</table>
### Table 4: The flow properties of MG1 and MG2

<table>
<thead>
<tr>
<th>Properties</th>
<th>MG1</th>
<th>MG2</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>0 days</td>
<td>10 days</td>
</tr>
<tr>
<td>Homogeneity rate (1/10mm)</td>
<td>60.54±1.6</td>
<td>99.58±2.9</td>
</tr>
<tr>
<td>Tensile strength (Kg/cm)</td>
<td>30.25±0.9</td>
<td>35.62±1.7</td>
</tr>
<tr>
<td>Elongation rate (%)</td>
<td>200.21±1.4</td>
<td>282.3±2.5</td>
</tr>
<tr>
<td>Spreadability(cm)</td>
<td>9.80±0.9</td>
<td>9.98±0.9</td>
</tr>
</tbody>
</table>

n=3, (±): sd

### Table 5: Optimized Parameter of HPTLC for quercetin isolation

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>Stationary Phase</td>
<td>Merck Silica gel 60 F254 HPTLC precoated plates</td>
</tr>
<tr>
<td>Mode of separation</td>
<td>Normal phase</td>
</tr>
<tr>
<td>Mobile phase</td>
<td>Ethyl acetate: Toluene: Formic acid (4:3.5:0.5v/v/v)</td>
</tr>
<tr>
<td>Chamber saturation</td>
<td>30min</td>
</tr>
<tr>
<td>Sample applicator</td>
<td>Camag Linomat V</td>
</tr>
<tr>
<td>Distance from the edges of the plat</td>
<td>13.0mm</td>
</tr>
<tr>
<td>Lamp and wavelength</td>
<td>Deuterium, 246nm</td>
</tr>
<tr>
<td>Densitometric scanner</td>
<td>Camag Scanner IV equipped with win-CATS Planar Chromatography manager software version 1.4.7</td>
</tr>
</tbody>
</table>

- Plate size: 4.0cm x 10.0cm
- Development chamber: Camag twin trough chamber
- Bandwidth: 7.0mm
- Space between the bands: 7.0mm
- Syringe: Hamilton, 100.0µL
- Rate of a sample application: 150nL/sec
- Development distance: 85.0mm
Table 6: Stability evaluation of transferosomal gel after 3 months

<table>
<thead>
<tr>
<th>Properties</th>
<th>Gel formulation</th>
<th>MG1</th>
<th>MG2</th>
</tr>
</thead>
<tbody>
<tr>
<td>k) Color</td>
<td>Slightly yellowish</td>
<td></td>
<td>Greenish</td>
</tr>
<tr>
<td>l) pH</td>
<td>5.62±0.23</td>
<td>5.89±0.95</td>
<td></td>
</tr>
<tr>
<td>m) Viscosity(CPS)</td>
<td>14005±230</td>
<td>13202±456</td>
<td></td>
</tr>
<tr>
<td>n) EE (%)</td>
<td>84.96±2.35</td>
<td>78.35±3.69</td>
<td></td>
</tr>
<tr>
<td>o) Drug content (%)</td>
<td>98.01±2.61</td>
<td>89.40±3.10</td>
<td></td>
</tr>
<tr>
<td>k) Color</td>
<td>Slightly yellowish</td>
<td></td>
<td>Greenish</td>
</tr>
<tr>
<td>l) pH</td>
<td>5.95±1.02</td>
<td>6.23±1.63</td>
<td></td>
</tr>
<tr>
<td>m) Viscosity(cps)</td>
<td>13750±412</td>
<td>12889±362</td>
<td></td>
</tr>
<tr>
<td>n) EE (%)</td>
<td>83.69±4.02</td>
<td>73.23±5.06</td>
<td></td>
</tr>
<tr>
<td>o) Drug content (%)</td>
<td>97.96±00.97</td>
<td>84.25±2.96</td>
<td></td>
</tr>
</tbody>
</table>

n=3, (±): sd

Legend
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*Ex vivo* percent drug release

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*uncorrected proof*
Figure 15: Dissolution profile of MG2 and MG2 gel

*Ex vivo* percent drug release

![Dissolution profile of MG2 and MG2 gel](image)